MATERIALS ENGINEERING AND NANOTECHNOLOGY (LM56)

(Lecce - Università degli Studi)

| Teaching CHEMISTRY 2 | | Teaching in italian CHEMISTRY 2 | Course year 1 |
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| • | | Teaching CHEMISTRY 2 | Language ENGLISH |
| GenCod A003109 Owner professor Giuseppe CICCARELLA | | SSD code CHIM/07 | Curriculum PERCORSO COMUNE |
| | | Reference course MATERIALS ENGINEERING AND | |
| | | Course type Laurea Magistrale | Location Lecce |
| | | Credits 9.0 | Semester First Semester |
| | | Teaching hours Front activity hours: 81.0 | Exam type Oral |
| | | | Assessment Final grade |
| | | For enrolled in 2019/2020 | Course timetable |
| | | Taught in 2019/2020 | https://easyroom.unisalento.it/Orario |
| BRIEF COURSE DESCRIPTION REQUIREMENTS | The purpose of this course is to introduce students to the molecular-level understanding of the physicochemical properties of organic substances aimed at characteristics of materials. General Chemistry | | |
| COURSE AIMS | Ability to manage organic chemistry issues. Ability to perform basic organic spectral analysis | | |
| TEACHING METHODOLOGY | Lecture-style instruction / Exercises and Practice Test Questions | | |
| ASSESSMENT TYPE | The exam consists of two parts: Part 1 - the student is asked to provide a full structural interpretation of FT-IR, MS, 1H- and 13C- NMR spectra and to elucidate the structures of an unknown compound (2 hours); Part 2 - the student is asked to illustrate two thehourstical topics; it is aimed to verify to what extent the student has gained knowledge and understanding of the selected topics of the course and is able to communicate about his understanding. | | |
| OTHER USEFUL INFORMATION | | appointment; contact the professor by erree Comprehensive Chemical Drawing P | email or at the end of class meetings. Tackage can be downloaded at the following |

URL: https://www.acdlabs.com/resources/freeware/chemsketch/



FULL SYLLABUS

Covalent bonds and shape of molecules (2 hours). Acids and bases (2 hours). Alkanes and Cycloalkanes (2 hours). Alkenes (2 hours). Alkenes: Reactivity (3 hours). Chirality (3 hours). Alkynes (2 hours). Alkyl halides (3 hours). Alcohols, ehters and thiols (1 hour). Benzene and its derivatives (3 hours). Amines (1 hour). Aldehydes and ketones (2 hours). Carboxylic acids (3 hours). Functional derivatives of carboxylic acids (3 hours). Infrared spectroscopy (6 hours). Mass Spectrometry (6 hours). NMR Spectroscopy (10 hours). Tutorials (27 hours)

REFERENCE TEXT BOOKS

William H. Brown, Thomas Poon, Introduction to Organic Chemistry, 6th Edition, Wiley